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The mass in grams/mole, equal to the sum of all the atomic mas... A formula that represents the simplest ratio among the element... Th unit of an ionic compound that represents the simplest rati... The type and actual number of atoms in a covalent compound. molar mass The mass in grams/mole, equal to the sum of all the atomic mas... empirical formula A...

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CHAPTER 5 REVIEW. The Periodic Law.  
SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered. (a) according to decreasing atomic mass. (b) according to Mendeleev's original design.

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a force that opposes motion between surfaces that are touching Asked in Animal Life , Ecosystems

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Answer Key ...

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Holt psychology chapter 5 vocab. The pattern of feelings, thoughts and behavior that sets peopl... An aspect of personality that is considered to be reasonably c... A person who tends to be more interested in his or her own tho... A person who tends to be active and self expressive and to gai... Personality The pattern of feelings,...

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Life Science Holt Ch. 1 A group of individuals that belong to the same

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species and liv... the number of individuals of the same species that live in a g... the pattern of distribution of organisms in a population.

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a Prokaryote [Holt Science and Technology (Cells: The Basic Units of Life) Section Review, Section: The Diversity of Cells] Asked in Science  
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## 6 Chemical Bonding

Concept Review j j. 1. A chemical is any substance that has [definite, indefinite] composition. Changes in chemicals, or chemical reactions, take place [only in test tubes, all] 2. The type and arrangement of [particles, crystals] in a sample of matter determine the properties of the matter.

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## Skills Worksheet Section Review

CHAPTER 6 REVIEW Chemical Bonding  
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a)



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formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) positive ions. (d) negative ions.

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Worksheets NAME \_\_\_\_\_ DATE \_\_\_\_\_

CLASS \_\_\_\_\_ The Science of Physics

Chapter 1 Mixed Review HOLT PHYSICS

1. Convert the following measurements to the units specified. a. 2.5 days to seconds b. 35 km to millimeters c. 43 cm to kilometers d. 22 mg to kilograms e. 671 kg to micrograms

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Answers to the Reading Checks are in the text's appendices so students can

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self-check. Chapters are broken down into sections. Each section concludes with a summary of key points, vocabulary activities, comprehension questions, a math application question, and critical thinking questions.

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